

Redox Cycling of o-Naphthoquinones in Trypanosomatids

SUPEROXIDE AND HYDROGEN PEROXIDE PRODUCTION

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ABSTRACT. β-Lapachone and structurally related lipophilic o-naphthoquinones, namely, CG 8-935, CG 9-442, CG 10-248 and mansonones A, C, E, and F, were investigated for redox cycling, production of reactive oxygen species, and cytotoxicity in the trypanosomatids Crithidia fasciculata and Leptomonas seymouri. Structural analysis of the assayed quinones indicated that a tricyclic structure, including a naphthalene ring, a 1,2b or 1,8bc pyran ring, and two ortho-carbonyl groups were required for quinone activities. The contribution of oxygen radical production to quinone cytotoxicity was supported by: (a) spectroscopic observation of quinone redox cycling; (b) production of the semiquinone radical; (c) H_2O_2 and O_2 production; (d) the effect of β-lapachone on thiol pools in C. fasciculata; (e) the effect of quinones on cell respiration; (f) superoxide dismutase inactivation after incubation of C. fasciculata with CG 8-935; and (g) the effect of quinones on cell growth. Copyright © 1996 Elsevier Science Inc. BIOCHEM PHARMACOL 52;12:1875–1882, 1996.

KEY WORDS. *o*-naphthoquinones; lapachones; mansonones; superoxide; hydrogen peroxide; oxidative damage; *Crithidia fasciculata*; *Leptomonas seymouri*.

Naphthoquinones are biologically important compounds, especially because of their pharmacological and cytotoxic action. The lipophilic *o*-naphthoquinone β-lapachone^{||} and related quinones (a) inhibit tumor cell growth [1–7]; (b) suppress HIV-1 replication in both acute and chronic infection [8]; (c) prevent oncogenic transformation [5]; (d) inhibit growth, DNA, RNA, and protein synthesis in *Try-panosoma cruzi*, the agent of Chagas' disease [9]; (e) produce strand breaks in the parasite DNA [9]; (f) induce produc-

tion of reactive oxygen species in *T. cruzi* and related trypanosomatids [10–16]; (g) induce chromosomal alterations [17]; (h) modify topoisomerase I activity [18]; (i) have potential clinical utility against human leukemia and prostate cancer [7]; and (j) inhibit microsomal electron transfer and lipid peroxidation [19].

To (a) establish structure-activity relationships, and (b) ascertain the role of oxygen radicals for o-naphthoquinone toxicity in trypanosomatids, we have compared in the present study the effects of a series of \(\beta-lapachone analogues, namely, CG 8-935, CG 9-442, CG 10-248, and mansonones A, C, E, and F, using as assay organisms the trypanosomatids Crithidia fasciculata and Leptomonas seymouri. These organisms are useful experimental systems for studying the factors involved in determining drug susceptibility [20-23]. Moreover, C. fasciculata and L. seymouri have the advantage of their relatively rapid growth and their lack of pathogenicity for humans. Catalase content was relatively high in C. fasciculata, a feature that allowed Gutteridge et al. [24] to examine this enzyme function in trypanosomatids. Figure 1 shows the structures of the quinones investigated, which included: (a) quinones with two carbonyls in C1 and C2, one pyranic oxygen in C4, and one 1,2b pyran ring (β-lapachone and CG quinones) [4]; (b) quinones with two carbonyls in C1 and C2, one pyranic oxygen in C4, and one 1,8bc pyran ring (mansonones E and F) [25]: (c) one quinone with two carbonyls in C1 and C2 (mansonone C) [25]; and (d) an o-tetrahydronaphthoguinone

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[&]quot;Abbreviations: GSH, reduced glutathione; SOD, superoxide dismutase; β-lapachone, 3,4-dihydro-2,2-dimethyl-2*H*-naphtho[1,2*b*]pyran-5,6-dione; α-lapachone, 3,4-dihydro-2,2-dimethyl-2*H*-naphtho[2,3*b*]-pyran-5,10-dione; CG 8-935, 3,4-dihydro-2-methyl-2-ethyl-2*H*-naphtho[1,2*b*]pyran-5,6-dione; CG 9-442, 3,4-dihydro-2-methyl-2-phenyl-2*H*-naphtho[1,2*b*]pyran-5,6-dione; CG 10-248, 3,4-dihydro-2,2-dimethyl-9-chloro-2*H*-naphtho[1,2*b*]pyran-5,6-dione; mansonone A,5,6,7,8-tetrahydro-3,8-dimethyl-5-isopropyl-1,2-naphthoquinone; mansonone C, 3,8-dimethyl-5-isopropyl-1,2-naphthoquinone; mansonone F, 3,6,9-trimethyl-naphtho[1,8-*b*c]pyran-7,8-dione; mansonone F, 3,6,9-trimethyl-naphtho[1,8-*b*c]pyran-7,8-dione; DMFA, dimethylformaide; MP-11, microperoxidase, the heme-portion of the cytochrome *c* molecule, with amino acids 11–21 still attached; and MOPS, 3-[*N*-morpholino]propanesulfonic acid.

FIG. 1. Structures of lipophilic o-naphthoquinones. (I) β-lapachone; (II) CG 8-935; (III) CG 9-442; (IV) CG 10-248; (V) mansonone A; (VI) mansonone C; (VII) mansonone E; (VIII) mansonone F; and (IX) α-lapachone.

with two carbonyls in C1 and C2 (mansonone A) [25]. For comparative purposes, α -lapachone, a p-naphthoquinone with two carbonyls in C1 and C4 and one pyranic oxygen in C2, was assayed.

MATERIALS AND METHODS Culture and Treatment of Trypanosomatids

C. fasciculata ATCC 11745 and the L. seymouri strain supplied by Dr. S. H. Hutner were used in most of the experiments reported here. C. fasciculata ATCC 12858 and L. seymouri ATCC 30220 were purchased from the American Type Culture Collection, Rockville, MD, U.S.A. C. fasciculata and L. Seymouri were grown in a culture medium containing (in g/L): NaCl, 5.0; KCl, 2.0; NaH₂PO₄, 0.5; sucrose, 2.5; sodium-ascorbate, 0.2; MOPS, 0.5; desiccated liver powder, 3.0; pluripeptone, 7.5; Brain-Heart infusion, 7.5; hemin [dissolved in 1:1 (v/v) triethanolamine:H₂O], 0.02. Cultures were incubated at 28° for 48 hr (C. fasciculata), or at 26° for 72 hr (L. seymouri) in a model G-25 New Brunswick Gyratory Shaker (New Brunswick Scientific Co., Inc., Edison, NJ, U.S.A.), at 100 rpm. Cell growth was measured by the increase of culture turbidity, at fixed incubation times. The cell concentration/turbidity ratio was determined using cell suspensions, whose concentration was predetermined by cell counting in a Neubauer chamber.

To assay quinone inhibition of trypanosomatid growth, quinones were dissolved in ethanol or DMFA, and aliquots of these solutions were added to duplicate 10-mL samples of sterile culture medium, before the inoculum. To the control samples, the corresponding volume of solvent was added. Incubations were performed in the New Brunswick Gyratory Shaker as indicated above. Three or more quinone

concentrations were assayed in each experiment and from the corresponding inhibition/concentration plots, the quinone concentration producing 50% inhibition of growth (IC_{50}) was calculated.

Cell Disruption and Homogenization

Cultured cells were collected by centrifugation at 750 g and 4°, for 10 min. The pellet was washed twice with 10–30 mL of 154 mM NaCl, and after centrifugation the cells were subjected to four cycles of freeze–thawing, using liquid nitrogen. The homogenate was centrifuged at 9800 g for 30 min, and the supernatant ("cell extract") was collected and used for enzyme measurements.

Analytical Procedures

Superoxide anion generation was measured by the adrenochrome method [26, 27]. The reaction mixture contained 20 mM KH₂PO₄-Na₂HPO₄, pH 7.4, 154 mM NaCl, 1.0 mM epinephrine, quinone, and cells as indicated under Results; total volume, 3.0 mL. Adrenochrome production was measured spectrophotometrically at 485–575 nm (ϵ = 2.95 mM⁻¹·cm⁻¹). Addition of 6 U/mL SOD verified that the absorbance increase was due to O₂-. H₂O₂ generation was measured by the MP-11 method [28]. The reaction mixture contained 20 mM KH₂PO₄-Na₂HPO₄, pH 7.4, 154 mM NaCl, 3.3 μ M MP-11, quinone and cells as indicated under Results; total volume, 3.0 mL. MP-11 absorbance was measured spectrophotometrically, at 419–407 nm (ϵ = 78 mM⁻¹·cm⁻¹). Addition of 60 U/mL catalase confirmed that the absorbance variation was due to H₂O₂.

Thiol compounds were determined using the ophthalaldehyde reaction [29]. Briefly, cell samples were collected by centrifugation and subjected to four freezethawing cycles, using liquid nitrogen. The homogenate was diluted with 0.1 M Na₂HPO₄, pH 8.0, 5.0 mM EDTA (phosphate-EDTA buffer). One milliliter of 25% (w/v) HPO₃ was added to 3.7 mL of cell extract, and the precipitated protein was discarded by centrifugation. The supernatant was diluted 10-fold with the phosphate-EDTA buffer, and the diluted samples were used for analysis. The assay mixture contained 0.1 mL of cell extract or GSH standard solution, 0.1 mL of o-phthalaldehyde solution (1 mg/mL methanol) and 1.8 mL of phosphate-EDTA buffer. After a 15-min incubation at 20-25° in the dark, fluorescence was measured at 350 nm (excitation) and 420 nm (emission) in an Aminco-Bauman (Urbana, IL, U.S.A.) spectrofluorometer.

Oxygen uptake was measured at 30° with a Gilson (Middleton, WI, U.S.A.) Oxygraph, model 5/6, using a Clark electrode. Spectrophotometric measurements were performed with an Aminco DW-2TM UV/VIS, or an Aminco Chance spectrophotometer.

HPLC analysis of quinones was performed using a Spherisorb, 5 μ m, 250 \times 4 mm reverse-phase C18 column (from Pharmacia-LKB, Piscataway, NJ, U.S.A.) attached to a

Pharmacia-LKB system constituted by UV Detector VWM 214, HPLC Pump 2248, and LKB 2221 Integrator. The mobile phase was methanol— H_2O (30:70, v/v), and the flow rate was 0.8 mL/min. Quinones were dissolved in the mobile phase (1 μ g/mL), and the injection volume was 10 μ L. The UV detector was set at wavelengths selected from the quinone ultraviolet spectra.

Enzyme Assays

Catalase activity was determined at $22-25^{\circ}$ by the rate of H_2O_2 decomposition, measured spectrophotometrically at 240 nm ($\varepsilon = 40 \text{ mM}^{-1} \cdot \text{cm}^{-1}$) [30]. The reaction mixtures contained cell extract, 20 mM KH₂PO₄-Na₂HPO₄, pH 7.4, and 154 mM NaCl. The reaction was started by adding 10 mM H_2O_2 . Activities were expressed in k U (sec⁻¹/ mL · (mg protein)⁻¹). SOD activity was determined at 30° by the inhibition of adrenochrome production [31], measured spectrophotometrically at 480 nm. The reaction mixture contained 1.0 mM epinephrine, 50 mM glycine-NaOH, pH 10.2, and 1.0 mM EDTA. The reaction was started by adding the cell extract. SOD activity is expressed in units, one unit being the amount of enzyme that inhibited by 50% the rate of epinephrine oxidation. Protein concentration was determined by the biuret method [32].

ESR

Measurements were performed with a Bruker (Bruker Analytische Messtechnik GMBH, Rheinstellen, Karlsruhe, Germany) ER 200 tt x-band ESR spectrometer equipped with a TE₁₀₂ cavity, at room temperature (28°). The general instrumental conditions were: microwave power, 21 mW; modulation frequency, 100 KHz; microwave frequency, 9.72 GHz; time constant, 0.5 sec; scan rate, 0.24 mT/min; gain, 1.0×10^6 . In all measurements the field was centered at 0.349 T. With C. fasciculata and CG 10-248, the assay mixture contained Crithidia (12.4 mg protein/mL) in 20 mM KH₂PO₄-Na₂HPO₄, pH 7.4, and 154 mM NaCl (gassed with nitrogen for 5 min in the spectrometer cell). The guinone pregassed with nitrogen was added (final quinone concentration, 2 mM), and signals were recorded with modulation amplitude of 0.04 mTpp. Controls without Crithidia or quinone were examined negatively for ESR.

Chemicals

β-Lapachone and α-lapachone were supplied by the Program for the Synthesis of Antiparasitic Drugs, Universidad Federal de Rio de Janeiro, Brazil; CG quinones were supplied by CIBA-GEIGY, Basel, Switzerland; and mansonones A, C, E, and F were isolated and supplied by Dr. C. Galeffi, Istituto Superiore di Sanitá, Rome, Italy. Quinols were prepared as described [33]. Other reagents were purchased from the following suppliers: antimycin A, D-glucose, sucrose, NaBH₄, MP-11 (prepared by enzymatic degradation of equine heart cytochrome c), SOD (from

bovine erythrocytes), glucose oxidase, catalase, EDTA, MOPS, bovine serum albumin, sodium ascorbate, sodium dithionite, hemin from the Sigma Chemical Co., St. Louis, MO, U.S.A.; DMFA from the T. Baker Chemical Co., Phillisburg, NJ, U.S.A.; argon from Matheson, East Rutherford, NJ, U.S.A.; Brain–Heart infusion and Bacto^R Desiccated Powdered Beef Liver, from Difco Laboratories, Detroit, MI, U.S.A.; Pluripeptone from Laboratorio Britania, Buenos Aires, Argentina.

o-Naphthoquinones are unstable [7] and degrade over time. Accordingly, the available compounds were examined for purity by HPLC. β-Lapachone, α-lapachone, CG quinones, and mansonone F were 100% pure; mansonones A, C, and E were 98.2, 98.4, and 97.3% pure, respectively.

Expression of Results

Unless otherwise indicated, means ± SEM are given. When values were the average of duplicate measurements, the experimental values deviated from the mean by < 5%. Statistical analysis was performed using ANOVA or Student's *t*-test. Assays were systematically repeated in C. Fasciculata and L. seymouri, but only typical experiments are illustrated.

RESULTS Quinone Redox Cycling

Quinone redox cycling in C. fasciculata and L. seymouri was demonstrated by absorption measurements at specific wavelengths. Figure 2 illustrates typical results obtained with CG 8-935, CG 9-442, and C. fasciculata at 450 nm. Experiments A and C show that (a) addition of quinone to the anaerobic cell suspension caused an initial increase of absorption, due to the quinone absorbance; (b) immediately thereafter, a rapid decrease followed, due to quinone reduction by cellular electron donors; (c) a relatively stable redox state was then reached, in which about 90% of the quinone was reduced; (d) oxygenation of the anaerobic cell suspension produced a rapid increase in absorption, especially with CG 8-935, indicating quinol oxidation; (e) a new redox state was then reached, in which about 23% of CG 8-935 or 47% or CG 9-442 remained reduced; (f) finally, sodium dithionite re-reduced the quinone to quinol. Omission of the anaerobic step (experiments B and D) resulted in a simpler reaction sequence in which, after several minutes of incubation, 93% of CG 8-935 or 78% of CG 9-442 remained in the oxidized state. Similar results were obtained with mansonones E and F (tracings not shown). Different kinetics were observed with mansonones A and C, as illustrated by a typical experiment with L. seymouri (Fig. 3). The kinetics of the anaerobic reduction of these mansonones resembled that of the CG quinones (Fig. 2) but, after oxygenation, the rate of quinol oxidation was negligible (with mansonone A) or very slow (with mansonone C). Moreover, reduction also occurred aerobically, in con-

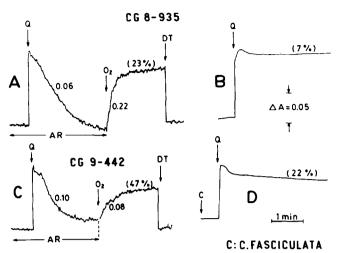


FIG. 2. CG 8-935 and CG 9-442 redox cycling in C. fasciculata. The reaction mixture contained Crithidia (0.96 mg/ mL), 20 mM KH₂PO₄-Na₂HPO₄, pH 7.4, 154 mM NaCl, 50 mM glucose, glucose oxidase (1.0 µg/mL) and catalase (283 U/mL); total volume, 3.0 mL. Experiments A and C: The reaction mixture and the quinone solution were pre-gassed with argon (AR) and continuously stirred with a magnetic bar at 30°. Q indicates the addition of 50 µM quinone; O2 indicates gassing with pure oxygen, and DT indicates addition of 1.2 mM sodium dithionite. The numbers near the tracings indicate rate of quinone reduction or quinol oxidation, in nmol/min/mg cell protein; values in parentheses indicate total quinol production, as a percentage of the initial quinone concentration. Experiments B and D: conditions were as in A and C, except for the gassings with argon and oxygen, which were omitted. Other experimental conditions were as described under Materials and Methods.

trast to the results obtained with the CG quinones (Fig. 2), experiments B and D).

Quinone reduction and quinol oxidation rates were calculated from the experiments described in Fig. 2 and 3, and similar ones were performed with other o-naphthoquinones. Table 1 shows that (a) with mansonones A and C, the rate of quinone reduction was greater than that of quinol oxidation; (b) with CG 8-935 and mansonones E and F, the rate of quinol oxidation was greater than the rate of quinone reduction; (c) with CG 9-442, the rates of quinone reduction and quinol oxidation were similar. The characteristic redox cycling of CG quinones and mansonones was confirmed by reduction with sodium dithionite, under air (spectra not shown).

Effect of Quinones on Cell Respiration

Table 2 shows the effect of quinones on C. fasciculata respiration. Quinones enhanced the rate of oxygen consumption, β -lapachone, CG quinones and mansonones E and F being the most effective. On the other hand, mansonone A and α -lapachone did not increase oxygen consumption above the control level, at least not to a significant degree. Quinone effects became more evident after inhibiting en-

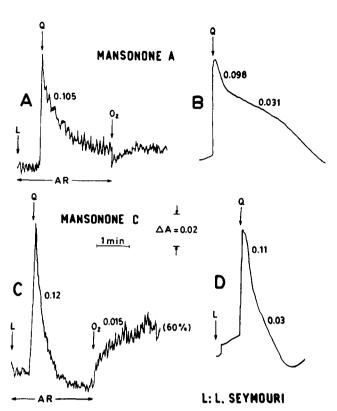


FIG. 3. Redox reactions of mansonones A and C in L seymouri. Experimental conditions were as in the legend of Fig. 2, except for the quinones. Quinone concentration, 50 μ M (Q).

dogenous respiration with antimycin A, which is known to act upon C. fasciculata mitochondrial electron transfer [34].

Effect of Quinones on Thiol Pools

C. fasciculata contains soluble thiol compounds, namely, GSH, glutathionyl-spermidine, dihydrotrypanothione [35], and ovothiol A [36]. Total soluble thiols, therefore, were considered as interesting markers of oxidative damage and, accordingly, worth measuring. The corresponding values

TABLE 1. Rate of quinone reduction and quinol oxidation in C. fasciculata and L. seymouri

	Reaction rate (nmol/min · (mg protein) ⁻¹)			
	C. fasciculata		L. seymouri	
Quinone (50 µM)	Quinone reduction	Quinol oxidation	Quinone reduction	Quinol oxidation
CG 9-442	93	80	100	110
CG 8-935	82	225	65	230
Mansonone A	151	10	96	16
Mansonone C	115	13	107	7
Mansonone E	43	1165	31	1950
Mansonone F	60	244	60	1950

Experimental conditions were as described in Fig. 2. Values are the averages of duplicate measurements.

TABLE 2. Effect of quinones on C. fasciculata respiration

Oxygen consumption

 $21.4 \pm 1.4 \dagger$

 4.0 ± 0.2

 4.0 ± 0.2

	(nmol O ₂ /min · (mg protein) ⁻¹) C. fasciculata		
Quinone (10 μM)	Antimycin omitted	Antimycin added	
None	17.1 ± 2.6	4.3 ± 0.5	
β-Lapachone	$36.1 \pm 2.1*$	$22.5 \pm 1.6 \dagger$	
CG 8-935	$35.7 \pm 2.2*$	$22.8 \pm 2.2 \dagger$	
CG 9-442	$33.7 \pm 3.0*$	$15.3 \pm 0.9 \dagger$	
CG 10-248	$36.4 \pm 2.3*$	$19.9 \pm 2.2 \dagger$	
Mansonone E	$33.1 \pm 2.0*$	$19.9 \pm 1.0 \dagger$	

The reaction mixture contained cells (0.5 to 0.7 mg protein/mL), 20 mM KH_2PO_4 - Na_2HPO_4 , pH 7.4, 154 mM NaCl, quinone, and antimycin (1 μ g/mL) as indicated above. Other experimental conditions were as described under Materials and Methods. Values are means \pm SEM (N=4).

32.1 ± 3.8*

 18.4 ± 1.9

 18.1 ± 1.6

Mansonone F Mansonone A

α-Lapachone

are presented as "XSH" in Table 3. The results obtained with β -lapachone and α -lapachone in C. fasciculata show that, after a 1-hr incubation, β -lapachone produced a significant decrease of XSH, whereas α -lapachone did not. Interestingly enough, glucose prevented quinone effects on XSH pools (Table 3). CG 9-442 and CG 10-248 behaved like β -lapachone (data not shown). Investigation of GSH-quinone adducts by HPLC yielded negative results (experimental data not shown).

Effect of Quinones on H₂O₂ and O₂⁻ Production

Quinone addition to C. fasciculata and L. seymouri enhanced $\rm H_2O_2$ and $\rm O_2^-$ production by these organisms. The effect of quinone was concentration dependent. Figure 4 shows the effect of CG 10-248 concentration on $\rm H_2O_2$ and $\rm O_2^-$ production. Table 4 summarizes the effect of quinones on $\rm H_2O_2$ and $\rm O_2^-$ production by L. seymouri and C. fasciculata. The results presented show that (a) β -lapachone,

TABLE 3. Effects of β -lapachone and α -lapachone on XSH in C. fasciculata

Quinone (10 µM)	Glucose (mM)	XSH (nmol/10 ⁸ cells)
None	0	22.2 ± 0.6
β-Lapachone	0	$13.4 \pm 0.6*$
α-Lapachone	0	24.0 ± 0.9
None	10	25.9 ± 0.2
β-Lapachone	10	20.8 ± 1.1
α-Lapachone	10	25.1 ± 0.3

The reaction mixture contained *Crithidia* (0.4×10^8 cells), 20 mM KH₂PO₄-Na₂HPO₄, pH 7.4, 154 mM NaCl, and additions as indicated above; total volume, 2.0 mL. Incubation was for 1 hr. Other conditions were as described under Materials and Methods. Values are means \pm SEM (N=3).

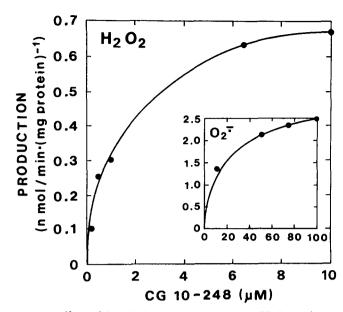


FIG. 4. Effect of CG 10-248 concentration on H₂O₂ and O₂⁻ production by L. seymouri. The reaction mixture contained Leptomonas (0.13 mg protein/mL), 20 mM KH₂PO₄-Na₂HPO₄, pH 7.4, 154 mM NaCl, and 3.3 µM MP-11 (H₂O₂ measurement) or 1.0 mM epinephrine (O₂⁻ measurement). Quinone concentration was as indicated on the abscissa. Other experimental conditions were as described under Materials and Methods. Values are the means of four or more experiments performed in duplicate.

CG quinones, and mansonones E and F enhanced H_2O_2 and O_2^- production, without significant differences between quinones; (b) the rate of O_2^- production was greater than that of H_2O_2 , in good agreement with the stoichiom-

TABLE 4. Effect of quinones on O_2^- and H_2O_2 production by C. fasciculata and L. seymouri

Quinone	O_2^- or H_2O_2 production (nmol/min · (mg protein) ⁻¹)		
(10 μM)	O ₂ -	H ₂ O ₂	
	Organism: C. fasciculata		
None	0.03 ± 0.01	0.05 ± 0.01	
CG 8-935	1.72 ± 0.51*	1.04 ± 0.04 *	
β-Lapachone	$1.22 \pm 0.10*$	$0.43 \pm 0.04*$	
Mansonone E	$2.20 \pm 0.20*$	0.48 ± 0.04*	
Mansonone F	$1.84 \pm 0.10*$	0.30 ± 0.14 *	
	Organism: L.	seymouri	
None	0.03 ± 0.01	0.06 ± 0.01	
CG 8-935	1.88 ± 0.82*	$0.35 \pm 0.08*$	
CG 10-248	$1.08 \pm 0.03*$	$0.65 \pm 0.02*$	
Mansonone E	1.80 ± 0.05*	$0.43 \pm 0.2*$	
Mansonone F	2.03 ± 0.43*	$0.23 \pm 0.05*$	

The reaction mixture contained cells (0.1 to 0.5 mg protein/mL), 20 mM KH $_2$ PO $_4$ · Na $_2$ HPO $_4$, pH 7.4, 154 mM NaCl, 1.0 mM epinephrine (O $_2$ ⁻ measurements) or 3.3 μ M MP-11 (H $_2$ O $_2$ measurements), and quinone as indicated above; final volume, 3.0 mL. The reaction was initiated by the addition of quinone. Other conditions were as described under Materials and Methods. Values are means \pm SEM (N \geq 3).

^{*}P < 0.003.

[†] P < 0.001.

^{*}P < 0.05.

^{*}P < 0.05

etry of the SOD reaction; (c) no significant differences were observed between H_2O_2 or O_2^- production rates by C. fasciculata and L. seymouri; (d) mansonone A did not enhance the rate of O_2^- and H_2O_2 production by the assayed organisms (experimental data not shown). Addition of catalase (60 U/mL) prevented the formation of the MP-11– H_2O_2 complex completely.

Production of O_2^- by the quinone-supplemented cell might involve quinol oxidation by oxygen. This hypothesis is supported by O_2^- production by quinol oxidation. Thus, with 25 μ M quinol in 20 mM potassium phosphate, pH 7.4, 154 mM NaCl, and 1.0 mM epinephrine, the rate of SOD (3.0 U/mL)-sensitive adrenochrome production was [in μ mol/min · (μ mol quinol)⁻¹]: 21.6 (CG 8-935), 19.0 (mansonone E), 0.01 (mansonone A), and 0.15 (mansonone C) (average of duplicate measurements). From the summarized results it may be concluded that CG 8-935 and mansonone E quinols were oxidized rapidly, producing O_2^- , whereas mansonone A and C quinols were not.

Semiquinone Production

The ESR spectrum of the CG 10-248 semi-quinone radical was observed after anaerobic incubation of the quinone with C. fasciculata. A similar spectrum was observed after CG 10-248 reduction with NaBH₄ (4.0 mg/mL; Crithidia omitted). The hyperfine splittings of 0.11, 0.13, and 0.15 mT indicated spin couplings at the protons at C5, C7, and C8 of the naphthalene ring. The semiquinone spectrum was confirmed by computational analysis using the coupling constants obtained experimentally (spectra not shown).

Effect of Quinones on Antioxidant Enzymes

Both *C. fasciculata* and *L. seymouri* contain catalase, but the enzyme activity in *Crithidia* is 5-fold greater than in *Leptomonas* [the specific activities in *C. fasciculata* and *L. seymouri* (k units per mg of protein) were (mean \pm SEM): 0.64 \pm 0.13 (N = 6) and 0.04 \pm 0.01 (N = 4), respectively]. Catalase and SOD may be inhibited by oxygen radicals and H_2O_2 [37–40]. Therefore, these enzymes were measured in *C. fasciculata* subjected to quinone-induced oxidative stress. The results in Table 5 indicate that only SOD activity decreased significantly, despite the relatively long incubation time and the high concentration of quinone. Under similar experimental conditions, trypanothione reductase was not modified (experimental data not shown).

Effect of Quinones on Trypanosomatid Growth

Table 6 summarizes the effect of quinones on the growth of C. fasciculata and L. seymouri. It is to be observed that both organisms were sensitive to the assayed quinones. β -Lapachone, CG quinones, and mansonones E and F were very effective, showing IC_{50} values in the 0.1 to $0.7~\mu M$ range. Mansonones A, C, and α -lapachone were less cytotoxic.

TABLE 5. Effect of CG 8-935 on catalase and SOD in C. fasciculata

	Enzyme activ	vity
CG 8-935 (μM)	SOD (U · (mg protein) ⁻¹)	Catalase (k)*
0 100	17.0 ± 3.1 7.5 ± 1.9†	0.64 ± 0.13 0.51 ± 0.17

C. fasciculata (73 × 10⁸ cells) were incubated in 20 mM KH₂PO₄- K_2 HPO₄, pH 7.4, 154 mM NaCl, containing CG 8-935 as indicated above; total volume, 10 mL. After 3 hr of incubation, cells were collected by centrifugation and homogenized, and enzyme activities in the supernatant were measured. Other experimental conditions were as described in Materials and Methods. Values are means \pm SEM (N \geq 3).

DISCUSSION

The biological activity of quinones (Tables 2, 3 and 6) allows one to establish major differences, with maximum activities for the CG quinones and mansonones E and F, and minimum activities for α -lapachone and mansonone A. These differences fit in well with the pro-oxidant activity of quinones, using dihydrolipoamide as reductant [41], and can be explained in terms of electronic symmetry and polarity [42]. Comparison of quinone activities with the rate of quinone reduction or quinol oxidation (Table 1 and text) indicates a better correlation with the latter parameter.

Taken together, our observations support the hypothesis that, in the assayed trypanosomatids, redox cycling and oxygen radical production contributed in a significant degree to o-naphthoquinone cytotoxicity. This conclusion is borne out by: (a) spectroscopic demonstration of quinone redox cycling (Fig. 2 and 3; Table 1); (b) H_2O_2 and O_2^- production (Fig. 4; Table 4); (c) the capability of quinols for O_2^- production (quoted in text); (d) the effect of quinones on trypanosomatid growth (Table 6); (e) semiqui-

TABLE 6. Effects of naphthoquinones on C. fasciculata and L. seymouri growth

	IС ₅₀ (μ M)
Quinone	C. fasciculata	L. seymouri
CG 8-935	0.79 ± 0.02	0.49 ± 0.06
CG 9-442	0.72 ± 0.14	0.64 ± 0.08
CG 10-248	0.76 ± 0.02	0.55 ± 0.08
α-Lapachone	27.0	4.1
Mansonone A	15.0	8.5
Mansonone C	6.0 (11.7)	4.0 (6.5)
Mansonone E	0.6 (0.9)	0.4 (0.4)
Mansonone F	0.3 (1.2)	0.1 (0.1)

C. fasciculata strain ATCC 11745, except for values in parentheses which were obtained with strain ATCC 12858. L. seymouri strain provided by Dr. S. H. Hutner, except for values in parentheses which were obtained with strain ATCC 30220. Other experimental conditions were as described in Materials and Methods. Values for CG 8-935, CG 9-442, adn CG 10-248 are presented as means \pm SEM (N \geq 3). Other values represent averages of duplicate 10_{50} determinations.

^{*} k, $sec^{-1}/mL \cdot (mg protein)^{-1}$.

[†] P < 0.05.

none radical production (quoted in text); (f) the effect of β -lapachone on thiol pools (Table 3); (g) the effect of quinones on cell respiration (Table 2); and (h) SOD inactivation (Table 5). The latter effect fits in well with SOD inactivation by H_2O_2 [37–40]. The variety of effects supports the role of "reactive oxygen species" in quinone cytotoxicity.

Catalase content in C. fasciculata was about 5-fold greater than in L. seymouri (see Results). Despite these differences, (a) the rates of quinone-dependent H₂O₂ production by these organisms were about the same (Table 4), and (b) quinones were equally cytotoxic to C. fasciculata and L. seymouri (Table 6). It seemed, therefore, that in the assayed organisms, catalase would not play a protective role against H₂O₂. This hypothesis agrees with the notion that, in trypanosomatids, catalase acts essentially as a heme store [24]. On the other hand, the discrepancy between the stimulated O_2 consumption and O_2^-/H_2O_2 production suggests that a large proportion of reactive oxygen species was scavenged by defense mechanisms, in close agreement with effects observed in several catalase- and peroxidasedeficient trypanosomatids [43–45]. In some parasites (T. cruzi), a non-enzymatic trypanothione-dependent peroxide metabolism has been detected [46].

Thiol oxidation and depletion by quinones may contribute to quinone cytotoxicity. In fact, intracellular thiol compounds, such as GSH, contribute to the regulation of DNA and protein synthesis and are essential cofactors for many enzymes [47, 48]. Interaction of quinones with a thiol compound would determine, among other effects, quinone reduction and one- or two-electron thiol oxidation [48–50]. The results in Table 3 fit in well with such an effect of quinones. On the other hand, the anti-oxidant effect of glucose in Table 3 may be explained by the coupling of the pentose phosphate cycle and the GSH redox cycle [51, 52].

DNA strand breakage seems to play an important role in quinone-induced cytotoxicity, as previously observed in T. cruzi exposed to low concentrations of β -lapachone, not α -lapachone [9]. It should be noted, however, that strand breakage and cell death induced by quinones can occur by mechanisms excluding oxygen radicals [53]. The very low IC50 values in Table 6 suggest that, in addition to oxygen radicals, other effects may be involved in quinone cytotoxicity, especially in actively dividing cells. Inactivation of topoisomerase I [18] may be one of these mechanisms. This enzyme inhibition would explain the almost equal activity of quinones on C. fasciculata and L. seymouri (Table 6).

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